Standard Free Energies of Formation

Substance	$\Delta G°_f$ (kJ/mol)	Substance	$\Delta G°_f$ (kJ/mol)	Substance	$\Delta G°_f$ (kJ/mol)
AgCl(*s*)	-109.8	$CaSO_4$(*s*)	-1334.3	N_2H_4(*g*)	+120.3
Al_2O_3(*s*)	-1582.2	Fe_2O_3(*s*)	-742.2	N_2H_4(*l*)	+110.5
$CHCl_3$(*g*)	-18.4	HBr(*g*)	-53.5	N_2O(*g*)	+104.2
CH_2Cl_2(*g*)	-68.9	HCl(*g*)	-95.3	N_2O_4(*g*)	+97.8
CH_2O(*g*)	-110.0	HF(*g*)	-324.4	N_2O_5(*g*)	+118.0
CH_3Cl(*g*)	-60.2	$HC_2H_3O_2$(*l*)	-328.6	NaCl(*s*)	-384.0
CH_3OH(*l*)	-166.1	H_2O(*g*)	-228.5	$NaHCO_3$(*s*)	-837.8
CH_4(*g*)	-50.8	H_2O(*l*)	-237.1	NaOH(*s*)	-379.6
CO(*g*)	-137.2	H_2O_2(*l*)	-120.3	Na_2CO_3(*s*)	-1048.0
CO_2(*g*)	-394.4	H_2S(*g*)	-33.3	O(*g*)	+238.1
C_2H_2(*g*)	+209.1	H_2SO_4(*l*)	-689.9	O_3(*g*)	+163.2
C_2H_4(*g*)	+68.4	KCl(*s*)	-408.8	PCl_3(*g*)	-269.6
C_2H_6(*g*)	-32.1	K_2SO_4(*s*)	-1319.6	PCl_5(*g*)	-208.6
C_2H_5OH(*l*)	-172.8	$MgCl_2$(*s*)	-592.0	PbO(*s*)	-189.3
$C_3H_5N_3O_9$(*l*)	unavailable	MgO(*s*)	-568.9	PbS(*s*)	-96.6
C_6H_6(*l*)	+124.3	$MgCO_3$(*s*)	-1002.3	SF_6(*g*)	-1116.4
$CaCO_3$(*s*)[1]	-1141.5	NH_3(*g*)	-16.4	SO_2(*g*)	-300.0
$CaCl_2$(*s*)	-748.1	NH_4Cl(*s*)	-203.1	SO_3(*g*)	-371.0
CaO(*s*)	-603.5	NO(*g*)	+86.6		
$Ca(OH)_2$(*s*)	-898.4	NO_2(*g*)	+51.3		

[1]calcite

All standard state, 25 °C and 1 bar (written to 1 decimal place).

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