Preclass Assignment CHEM 1100-General Chemistry II

Name:

Section: 31, TR

Due Date: Tuesday 3/31/2020

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1. For the titration of 25.00 mL of a 0.200 M HCl solution with 0.200 M NaOH, calculate the pH when 24.90 mL of base have been added. This is step F in the strong acid-strong base titation.

2. For the titration of 25.00 mL of a 0.200 M $HC_2H_3O_2$ solution with 0.200 M NaOH, calculate the pH when 24.90 mL of base have been added. This is step F in the weak acid-strong base titation.